

# Polybrominated diphenyl ethers from a sponge of the *Dysidea* genus that inhibit Tie2 kinase

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**Abstract**—Tie2 kinase, an enzyme that supports angiogenesis essential for tumor growth and survival, was selected as a target in a search for naturally occurring inhibitors of potential utility for antitumor therapy. Two polybrominated diphenyl ethers, 3,5-dibromo-2-(2',4'-dibromophenoxy)phenol (**1**) and 4,6-dibromo-2-(2',4'-dibromophenoxy)phenol (**2**) were isolated from an extract prepared from *Dysidea* sp. after bioassay-guided fractionation.

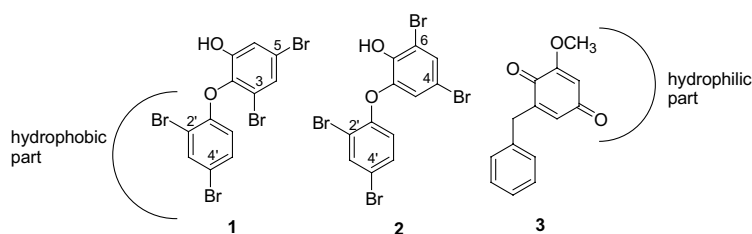
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## 1. Introduction

It has been shown that the growth and metastasis of solid tumors is dependent on tumor angiogenesis.<sup>1</sup> For several types of solid tumors, clinical studies have suggested that there is a direct correlation between the density of tumor vessels and an adverse prognosis in patients. These include breast, colon, lung, kidney, bladder, and head and neck cancers.<sup>2</sup> The Tie2 pathway and the VEGF receptor pathway have been shown to be two independent mediators essential for angiogenesis in vivo.<sup>3</sup> Blocking Tie2 kinase activation by i.v. administration of a recombinant, soluble Tie2 receptor was shown to inhibit tumor growth and metastasis.<sup>4</sup> These findings demonstrated a role for the Tie2 pathway in the formation of the tumor vascu-

lature and suggested that targeting proteins in the Tie2 kinase pathway might afford agents useful in anti-cancer therapy. Several Tie2 inhibitors are in clinical trials.<sup>5</sup>

A crude extract prepared from a sponge of the *Dysidea* genus was found to inhibit Tie2 kinase with an IC<sub>50</sub> of 6.67 µg/mL. Previous studies of the chemical constituents of sponges in this genus have resulted in the isolation of structurally diverse secondary metabolites including bromophenols,<sup>6</sup> sesquiterpenes,<sup>7</sup> sesterterpenes,<sup>8</sup> sterols,<sup>9</sup> and polychlorinated compounds.<sup>10</sup> Although many biological activities have been reported for these compounds, no Tie2 kinase inhibitory activity has been reported previously for the isolated compounds. Reported herein is the isolation of two



**Figure 1.** The structures of Tie2 inhibitors.

**Keywords:** Tie2 kinase inhibitor; Polybrominated diphenyl ethers; *Dysidea*; Sponge.

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polybrominated diphenyl ethers from the crude extract that inhibit Tie2 kinase (Fig. 1).

## 2. Results and discussion

The 1:1 CH<sub>2</sub>Cl<sub>2</sub>–CH<sub>3</sub>OH crude extract prepared from *Dysidea* sp. was applied to a diol column. The column was washed successively with hexanes, 1:1 hexanes–CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 1:1 CH<sub>2</sub>Cl<sub>2</sub>–CH<sub>3</sub>OH and CH<sub>3</sub>OH. The CH<sub>2</sub>Cl<sub>2</sub> fraction exhibited the strongest inhibition of Tie2 kinase. Crystallization from CH<sub>3</sub>OH–hexanes afforded active compound **1** as colorless flakes. The 1:1 hexanes–CH<sub>2</sub>Cl<sub>2</sub> fraction was fractionated further on a diol column. The column was washed successively with hexanes, 4:1 hexanes–CH<sub>2</sub>Cl<sub>2</sub>, 3:2 hexanes–CH<sub>2</sub>Cl<sub>2</sub> and CH<sub>2</sub>Cl<sub>2</sub>. The 3:2 hexanes–CH<sub>2</sub>Cl<sub>2</sub> fraction afforded active compound **2**.

The negative ESI-MS of **1** and **2** showed the same [M–1]<sup>–</sup> peak at *m/z* 501, but their <sup>1</sup>H NMR spectra were clearly different. Careful analysis of chemical shifts and coupling constants revealed that the two compounds had the same proton coupling pattern, but exhibited different chemical shifts. Their <sup>1</sup>H NMR spectra in DMSO-*d*<sub>6</sub> were same as 3,5-dibromo-2-(2',4'-dibromophenoxy)phenol (**1**)<sup>6b,11</sup> and 4,6-dibromo-2-(2',4'-dibromophenoxy)phenol (**2**)<sup>6b,11</sup> respectively (see Section 3). The <sup>13</sup>C NMR data (see Section 3) and X-ray diffraction measurement of the single crystal of **1** also supported the structure of **1**; therefore, the structure was unambiguously determined.

The two polybrominated diphenyl ethers were reasonably potent inhibitors of Tie2 kinase (Table 1) exhibiting IC<sub>50</sub> values of 2.1 and 6.2 μM for **1** and **2**, respectively. Other biological activities for compounds of this type have been noted, including antibacterial and fungicidal activity,<sup>6c</sup> brine shrimp toxicity,<sup>6c</sup> enzyme inhibition of inosine monophosphate dehydrogenase, guanosine monophosphate synthetase, and 15-lipoxygenase,<sup>6b</sup> inhibition of the cell division of fertilized sea urchin eggs,<sup>6e</sup> and cytotoxicity.<sup>6d</sup> Compound **1** has also been reported to inhibit contractile activity in the guinea pig ileum.<sup>12</sup>

Tie2 kinase is a receptor tyrosine kinase. Waldmann and co-workers<sup>13</sup> have described Tie2 kinase inhibitor **3**, which has a structure not dissimilar to those of polybrominated diphenyl ethers **1** and **2**. It was suggested that a hydrophobic moiety, which may interact with a hydrophobic pocket close to the binding site, and a hydrophilic moiety, which may participate in hydrogen bonding to the ATP binding site of kinases are necessary for inhibitory activity. It has been suggested<sup>13</sup> that a quinone flanked by a hydrophobic group may be important

for binding, but the present results indicate that a brominated phenol system may bind more efficiently to the ATP binding site because the diphenyl ethers **1** and **2** showed better Tie2 kinase inhibitory activity than quinone **3**. The discovery of the Tie2 kinase inhibitory activity of the two polybrominated diphenyl ethers provide good lead structures for further modification to afford improved Tie2 kinase inhibitors.

## 3. Experimental

### 3.1. General experimental procedures

Lichroprep diol (40–63 μm) is a product of EM Industries, Inc. <sup>1</sup>H NMR were measured on General Electric QE 300, GN-300 NMR spectrometers or a Varian Unity INOVA-500 spectrometer. Mass spectra were recorded on a Finnigan MAT 4600 mass spectrometer.

### 3.2. Bioassay

The bioassay for Tie2 kinase inhibitory activity was carried out by methods similar to those described previously.<sup>14</sup>

### 3.3. Fractionation and structure identification

The crude extract was prepared from a sponge of *Dysidea* sp. by soaking with 1:1 CH<sub>2</sub>Cl<sub>2</sub>–CH<sub>3</sub>OH. The crude extract (42 mg, IC<sub>50</sub> 6.67 μg/mL) was applied to a 20 g diol column. The column was washed successively with hexanes, 1:1 hexanes–CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 1:1 CH<sub>2</sub>Cl<sub>2</sub>–CH<sub>3</sub>OH and CH<sub>3</sub>OH. The 1:1 hexanes–CH<sub>2</sub>Cl<sub>2</sub> fraction (2.8 mg, IC<sub>50</sub> 10.75 μg/mL), CH<sub>2</sub>Cl<sub>2</sub> fraction (30.1 mg, IC<sub>50</sub> 6.67 μg/mL), and 1:1 CH<sub>2</sub>Cl<sub>2</sub>–CH<sub>3</sub>OH fraction (5.7 mg, IC<sub>50</sub> 3.14 μg/mL) exhibited strong inhibition of Tie2 kinase. The CH<sub>2</sub>Cl<sub>2</sub> fraction (20 mg) was crystallized from CH<sub>3</sub>OH–hexanes to afford compound **1** as colorless flakes (5.5 mg, IC<sub>50</sub> 1.06 μg/mL) and its mother liquor (14.3 mg, IC<sub>50</sub> 4.66 μg/mL). TLC analysis revealed that the CH<sub>2</sub>Cl<sub>2</sub>–CH<sub>3</sub>OH fraction also consisted mainly of **1**. The 1:1 hexanes–CH<sub>2</sub>Cl<sub>2</sub> fraction (2.8 mg) was fractionated further on a 10 g diol column. The column was washed successively with hexanes, 4:1 hexanes–CH<sub>2</sub>Cl<sub>2</sub>, 3:2 hexanes–CH<sub>2</sub>Cl<sub>2</sub>, and CH<sub>2</sub>Cl<sub>2</sub>. The 3:2 hexanes–CH<sub>2</sub>Cl<sub>2</sub> fraction afforded compound **2** (0.3 mg, IC<sub>50</sub> 3.33 μg/mL).

Compound **1**: <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 300 MHz): δ 6.42 (1H, d, *J* = 8.9 Hz, H-6'), 7.12 (1H, d, *J* = 2.3 Hz, H-6), 7.36 (1H, dd, *J* = 8.9, 2.3 Hz, H-5'), 7.37 (1H, d, *J* = 2.3 Hz, H-4), 7.85 (1H, d, *J* = 2.3 Hz, H-3'), <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz): δ 113.4 (C-2'), 116.6 (C-6), 116.8 (C-3), 118.4 (C-4'), 120.3 (C-6'), 120.5 (C-5), 128.5 (C-4), 132.2 (C-5'), 136.9 (C-3'), 139.3 (C-2), 150.9 (C-1), and 152.7 (C-1'); negative ion ESI-MS, *m/z* 1002 [2M–2]<sup>–</sup> and 501 [M–1]<sup>–</sup>. The above spectral data were same as those reported previously for compound **1**.<sup>6b,11</sup>

Compound **2**: <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 300 MHz): δ 6.90 (1H, d, *J* = 8.8 Hz, H-6'), 6.91 (1H, d, *J* = 2.3 Hz, H-3),

**Table 1.** IC<sub>50</sub> values of compounds **1–3** as inhibitors of Tie2 kinase

Compound	IC <sub>50</sub> (μM)
<b>1</b>	2.1
<b>2</b>	6.2
<b>3</b>	18 <sup>a</sup>

<sup>a</sup> Data from Ref. 13a.

7.51 (1H, d,  $J = 2.3$  Hz, H-5), 7.52 (1H, dd,  $J = 8.8$ , 2.3 Hz, H-5'), 7.93 (1H, d,  $J = 2.3$  Hz, H-3'); negative ion ESI-MS  $m/z$  1002  $[2M-2]^-$  and 501  $[M-1]^-$ . These spectral data were same as those reported previously for compound **2**.<sup>6b,11</sup>

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